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APPLICATION NO.	FILING DATE	FIRST NAMED INVENTOR	ATTORNEY DOCKET NO.	CONFIRMATION NO.
10/018,868	03/07/2002	Michel Devic	2988-694	8327
75	90 04/22/2004		EXAMINER	
Thomas F. Roland, Esq.			LANGEL, WAYNE A	
ATOFINA Chemicals, Inc. 2000 Market Street			ART UNIT	PAPER NUMBER
Philadelphia, PA 19103			1754	
		DATE MAILED: 04/22/2004		

Please find below and/or attached an Office communication concerning this application or proceeding.



UNITED STATES DEPARTMENT OF COMMERCE Patent and Trademark Office

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ATTORNEY DOCKET NO. FIRST NAMED INVENTOR FILING DATE SERIAL NUMBER EXAMINER ART UNIT PAPER NUMBER DATE MAILED: This is a communication from the examiner in charge of your application. COMMISSIONER OF PATENTS AND TRADEMARKS Responsive to communication filed on _____ This action is made final. This application has been examined A shortened statutory period for response to this action is set to expire ______ month(s), ______days from the date of this letter. Failure to respond within the period for response will cause the application to become abandoned. 35 U.S.C. 133 Part I THE FOLLOWING ATTACHMENT(S) ARE PART OF THIS ACTION: 2. Notice of Draftsman's Patent Drawing Review, PTO-948. Notice of References Cited by Examiner, PTO-892. 4. Notice of Informal Patent Application, PTO-152.

6. ______ Notice of Art Cited by Applicant, PTO-1449. Information on How to Effect Drawing Changes, PTO-1474... Part II SUMMARY OF ACTION _____ are pending in the application. 1. Claims_ 2. Claims_ 3. Claims_ 5. Claims _____ are subject to restriction or election requirement. 6. Claims 7. This application has been filed with informal drawings under 37 C.F.R. 1.85 which are acceptable for examination purposes. 8. Formal drawings are required in response to this Office action. . Under 37 C.F.R. 1.84 these drawings 9. The corrected or substitute drawings have been received on _ are ☐ acceptable; ☐ not acceptable (see explanation or Notice of Draftsman's Patent Drawing Review, PTO-948). _____. has (have) been approved by the 10. The proposed additional or substitute sheet(s) of drawings, filed on examiner; $\ \square$ disapproved by the examiner (see explanation). has been approved; disapproved (see explanation). 11. The proposed drawing correction, filed ____ 12. Acknowledgement is made of the claim for priority under 35 U.S.C. 119. The certified copy has been received on the claim for priority under 35 U.S.C. 119. ☐ been filed in parent application, serial no. ______; filed on _____ 13. Since this application apppears to be in condition for allowance except for formal matters, prosecution as to the merits is closed in

accordance with the practice under Ex parte Quayle, 1935 C.D. 11; 453 O.G. 213.

14. Other

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The following is a quotation of the appropriate paragraphs of 35 U.S.C. 102 that form the basis for the rejections under this section made in this Office action:

A person shall be entitled to a patent unless -

- (b) the invention was patented or described in a printed publication in this or a foreign country or in public use or on sale in this country, more than one year prior to the date of application for patent in the United States.
- (e) the invention was described in (1) an application for patent, published under section 122(b), by another filed in the United States before the invention by the applicant for patent or (2) a patent granted on an application for patent by another filed in the United States before the invention by the applicant for patent, except that an international application filed under the treaty defined in section 351(a) shall have the effects for purposes of this subsection of an application filed in the United States only if the international application designated the United States and was published under Article 21(2) of such treaty in the English language.

The following is a quotation of 35 U.S.C. 103(a) which forms the basis for all obviousness rejections set forth in this Office action:

(a) A patent may not be obtained though the invention is not identically disclosed or described as set forth in section 102 of this title, if the differences between the subject matter sought to be patented and the prior art are such that the subject matter as a whole would have been obvious at the time the invention was made to a person having ordinary skill in the art to which said subject matter pertains. Patentability shall not be negatived by the manner in which the invention was made.

Claims 32-41 are rejected under 35 U.S.C. 102(e) as anticipated by or, in the alternative, under 35 U.S.C. 103(a) as obvious over Zhou et al. No distinction is seen between the catalyst disclosed by Zhou et al, and that recited in applicant's claims.

Zhou et al disclose at col. 10, line 65 to col. 11, line5 that the noble metal particles have a size of 1-100 nm, and suggest at col. 5, lines 29-34 that the particles are present as clusters. Regarding claim 41, it is noted that the catalyst of Zhou et al is for the production of hydrogen production.

Claims 32-40 are rejected under 35 U.S.C. 102(b) as anticipated by or, in the alternative, under 35 U.S.C. 103(a) as obvious over Birbara et al. No distinction is seen between the catalyst disclosed by Birbara et al, and that recited in applicant's claims.

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Birbara et al disclose at col. 3, line 58 to col. 4, line 3 that that the noble metal crystallites have a size of about 10 angstroms or smaller. Tha crystallites would inherently be present as clusters, since Birbara et al disclose at col. 3, lines 8-13 and at col. 5, lines 17-26 that the particles are formed by reducing a noble metal halide.

Claims 15-41 are rejected under 35 U.S.C. 102(e) as anticipated by or, in the alternative, under 35 U.S.C. 103(a) as obvious over Sellin et al. No distinction is seen between the catalyst and method disclose by Sellin et al, and that recited in applicant's claims. The process of Sellin et al entails reducing a noble metal salt on a catalyst support (see col. 2, lines 18-50), and further disclose in the paragraph bridging columns 5 and 6 that the reaction medium for reacting hydrogen and oxygen to produce hydrogen peroxide contains bromide ions. Accordingly all the method steps recited in applicant's claim 15 are met. Such reaction medium is acidic.

Gosser et al and Bertsch-Frank et al are made of record for disclosing methods for preparing hydrogen peroxide by reacting hydrogen and oxygen in the presence of a noble metal catalyst.

Any inquiry concerning this communication should be directed to Wayne Langel at telephone number 571-272-1353.

> Wavne Langel Primary Examiner - Art Unit 1754

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